organic papers

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

Masataka Watanabe,^a Taisuke Matsumoto,^b Shuntaro Mataka^b and Thies Thiemann^b*

^aInterdisciplinary Graduate School of Engineering, Kyushu University, 6-1 Kasuga-koh-en, Kasuga-shi, Fukuoka 816-8580, Japan, and ^bInstitute of Materials Chemistry and Engineering, Kyushu University, 6-1 Kasuga-koh-en, Kasuga-shi, Fukuoka 816-8580, Japan

Correspondence e-mail: thies@cm.kyushu-u.ac.jp

Key indicators

Single-crystal X-ray study T = 123 KMean σ (C–C) = 0.003 Å R factor = 0.023 wR factor = 0.061 Data-to-parameter ratio = 15.2

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

17-Bromo-16-formyl-3-methoxyestra-1,3,5(10),16tetraene

In the title estrane derivative, $C_{20}H_{23}BrO_2$, the five-membered ring adopts a conformation that is intermediate between twist and envelope. The C atom of the formyl group and the bromo function are in close proximity. Close intermolecular $Br \cdots Br$ and $Br \cdots C$ contacts are observed in the crystal structure.

Comment

The title compound, (I), was synthesized by Thiemann *et al.* (1997) as an important precursor for steroidal bromobutadiene systems and *E*-ring annelated estranes (Thiemann *et al.*, 2001). The chloro analog of (I) is known to have a cholesterol-lowering effect (Kaneko & Yamato, 1969). We have used (I) in numerous metal-catalyzed cross-coupling processes, such as in Suzuki and Sonogashira reactions, with varying degrees of success (Watanabe *et al.*, 2005).



Steric restrictions in the transition state seem to have an influence on the success of the reactions. In order to understand the steric characteristics of (I), particularly the effects of the substituents on the five-membered ring (Fig. 1), a single-crystal X-ray diffraction study of (I) was carried out.



Figure 1

A view of the molecule of (I), with the atom-numbering scheme. Displacement ellipsoids are drawn at the 50% probability level; H atoms are shown as small spheres of arbitrary radii.

© 2005 International Union of Crystallography Printed in Great Britain – all rights reserved Received 13 January 2005

Accepted 7 March 2005

Online 18 March 2005



A view of the unit cell of (I) down the [212] direction. The intermolecular Br. . . Br and Br. . . C contacts are shown as dashed lines. [Symmetry codes: (i) $-x, y - \frac{1}{2}, 1 - z$; (ii) $-x, \frac{1}{2} + y, 1 - z$.]

There is one independent molecule of (I) in the asymmetric unit. Ring A shows little distortion from planarity, as is usual in other estrones and estradiols. The ring conformations in the molecule of (I) were analyzed using the Cremer & Pople (1975) puckering parameters. Ring B has a half-chair conformation [the Cremer & Pople puckering parameters are Q =0.496 (3) Å, $\theta = 4.61$ (3)° and $\varphi = 165.6$ (3)°]. This conformation is also in accordance with the endocyclic torsion angles within ring B (see Boeyens, 1978). Ring C, with a trans-fusion to rings B and D, has a chair conformation [Q = 0.590 (2) Å, $\theta = 6.7 (2)^{\circ}$ and $\varphi = 297 (2)^{\circ}$]. Ring D has a conformation between half-chair (twist) and envelope [Q = 0.383 (3) Å and] $\varphi = 206.8 \, (4)^\circ$, 51% half-chair (twist) character and 49% envelope character (Evans & Boevens, 1989) with a pseudorotation angle $P = 7.1 (2)^\circ$, and an upper limit of the endocyclic torsion angle $\tau_{\rm m}$ = 38.5 (1)° (Altona & Sundaralingam, 1972; Rao et al., 1981) for the atom sequence C13-C17].

The intramolecular distance from Br1 to C20 is 3.349 (2) Å, which is smaller than the sum of the van der Waals radii of C (1.70 Å) and Br (1.85 Å) [3.55 Å, according to Bondi (1964)]. These steric constraints are reflected in the poor results of certain transformations, such as the replacement of a bromoby an iodo-functionality at C17 of (I) using a nickel catalyst (Takagi et al., 1978). Classical Pd-catalyzed C-C bondforming reactions (de Meijere & Diederich, 2004) of (I), such as Suzuki cross-coupling, Sonogashira, and Heck reactions, work quite well (Watanabe et al., 2005). It can be deduced from the close proximity of Br1 and C20 that, after oxidative insertion of the palladium species into C17-Br1, with concomitant rotation about the C16-C20 bond, the geometry favors a stabilization of the Pd^{II} complex of (I) by the carbonyl O atom.

In the crystal structure, (I) packs in chains, each consisting of two rows of molecules, where parallel chains are arranged in a stepwise fashion. The dihedral angle between the mean plane of the steroidal frameworks of one row and the mean plane of the molecules in the second row is $73.90 (4)^{\circ}$. Between molecules of the two rows making up one chain, there exist halogen-carbon and halogen-halogen interactions

(Fig. 2), where the intermolecular $Br1 \cdots C20^{i}$ and $Br1 \cdots Br1^{i}$ distances are 3.508 (2) and 3.6165 (6) Å, respectively [symmetry code: (i) -x, $y - \frac{1}{2}$, 1 - z]. The Br1ⁱ \cdots Br1 \cdots C20ⁱ angle is 56.06 (4)°. Many C-halogen $\cdots \pi$ intermolecular interactions have been reported and have been studied primarily because of their importance in drug-protein interactions (Prasanna & Row, 2000). The Br1···C20ⁱ distance [3.508 (2) Å] in (I) is shorter than the mean value for the $Br \cdot \cdot \cdot C$ (intermolecular) distances reported in the Cambridge Structural Database (2004 Release; Allen, 2002) and the Bookhaven Protein Data Bank according to the study performed by Prasanna & Row (2000) [3.625 (9) Å for $Br \cdot \cdot \cdot C$ (intermolecular) for 387 such short contacts reported before 2000]. The distance $Br1 \cdots Br1^{i}$ in (I) is in the typical range of halogen-halogen interactions (3.2-4.0 Å for Cl and Br; Saruma & Desiraju, 1986, Moorthy et al., 2002).

Experimental

Compound (1) was prepared from 3-methoxyestra-1,3,5(10)-trien-17one (3-O-methylestrone) by an Arnold Vilsmeier reaction with PBr₃-DMF reagent (i.e. bromoiminium salt) according to the method described by Thiemann et al. (1997).

Crystal data

$C_{20}H_{23}BrO_2$	$D_x = 1.467 \text{ Mg m}^{-3}$
$M_r = 375.30$	Mo $K\alpha$ radiation
Monoclinic, P2 ₁	Cell parameters from 3041
a = 9.702 (3) Å	reflections
b = 6.093 (2) Å	$\theta = 3.6-27.5^{\circ}$
c = 14.640 (5) Å	$\mu = 2.43 \text{ mm}^{-1}$
$\beta = 100.914 \ (1)^{\circ}$	T = 123.1 K
$V = 849.7 (4) \text{ Å}^3$	Platelet, colorless
Z = 2	0.20 \times 0.18 \times 0.06 mm

Data collection

Rigaku Saturn diffractometer	3354 reflections with $I > 2\sigma(I)$
ω scans	$R_{\rm int} = 0.025$
Absorption correction: multi-scan	$\theta_{\rm max} = 27.5^{\circ}$
(Jacobson, 1998)	$h = -11 \rightarrow 12$
$T_{\min} = 0.711, \ T_{\max} = 0.864$	$k = -7 \rightarrow 7$
6860 measured reflections	$l = -19 \rightarrow 18$
3534 independent reflections	
-	

Refinement

Refinement on F^2	$(\Delta/\sigma)_{\rm max} < 0.001$
$R[F^2 > 2\sigma(F^2)] = 0.023$	$\Delta \rho_{\rm max} = 0.56 \ {\rm e} \ {\rm \AA}^{-3}$
$wR(F^2) = 0.061$	$\Delta \rho_{\rm min} = -0.65 \text{ e } \text{\AA}^{-3}$
S = 1.01	Extinction correction: none
3534 reflections	Absolute structure: (Flack, 1983),
232 parameters	1422 Friedel pairs
H-atom parameters constrained	Flack parameter: 0.000 (7)
$w = 1/[0.0001F_{\rm o}^2 + \sigma(F_{\rm o}^2) + 1.7]/$	
(AF^2)	

All H atoms were included in the refinement as riding on their parent atoms with $U_{iso}(H)$ values set to $1.2U_{eq}$ of the parent C atoms (C-H = 0.95 Å).

Data collection: CrystalClear (Rigaku Corporation, 1999); cell refinement: CrystalClear; data reduction: CrystalStructure (Rigaku/ MSC, 2004); program(s) used to solve structure: SIR97 (Altomare et al., 1999); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: PLATON (Spek, 2003); software used to prepare material for publication: CrystalStructure.

References

- Allen, F. H. (2002). Acta Cryst. B58, 380-388.
- Altomare, A., Burla, M., Camalli, M., Cascarano, G., Giacovazzo, C., Guagliardi, A., Moliterni, A., Polidori, G. & Spagna, R. (1999). J. Appl. Cryst. 32, 115–119.
- Altona, C. & Sundaralingam, M. (1972). J. Am. Chem. Soc. 94, 8205-8212.
- Boeyens, J. C. A. (1978). J. Cryst. Mol. Struct. 8, 317-320.
- Bondi, A. (1964). J. Phys. Chem. 68, 441-451.
- Cremer, D. & Pople, J. A. (1975). J. Am. Chem. Soc. 97, 1354-1358.
- Evans, D. G. & Boeyens, J. C. A. (1989). Acta Cryst. B45, 581-590.
- Flack, H. D. (1983). Acta Cryst. A39, 876-881.
- Jacobson, R. (1998). *REQAB*. Private communication to Rigaku Corporation. Kaneko, H. & Yamato, Y. (1969). Japanese Patent No. 44004345.
- Meijere, A. de & Diederich, F. (2004). Metal Catalysed Cross-coupling Reactions. Weinheim, Germany: VCH-Wiley.

- Moorthy, J. N., Natarajan, R., Mal, P. & Venugopalan, P. (2002). J. Am. Chem. Soc. 124, 6530–6531.
- Prasanna, M. D. & Row, T. N. G. (2000). Cryst. Eng. 3, 135-154.
- Rao, S. T., Westhof, E. & Sundaralingam, M. (1981). Acta Cryst. A37, 421–425. Rigaku Corporation (1999). CRYSTALCLEAR. Rigaku Corporation, Tokyo,
- Japan. Rigaku/MSC (2004). *CrystalStructure*. Version 3.6.0. Rigaku/MSC, 9009 New Trails Drive The Woodlands, TX 77381-5209, USA.
- Saruma, J. A. R. & Desiraju, G. R. (1986). Acc. Chem. Res. 19, 222–228.
- Sheldrick, G. M. (1997). SHELXL97. University of Göttingen, Germany.
- Spek, A. L. (2003). J. Appl. Cryst. **36**, 7–13.
- Takagi, K., Hayama, N. & Inokawa, S. (1978). *Chem. Lett.* pp. 1435–1436.
- Thiemann, T., Thiemann, C., Sasaki, S., Vill, V., Mataka, S. & Tashiro, M. (1997). J. Chem. Res. (S), pp. 248–249.
- Thiemann, T., Watanabe, M. & Mataka, S. (2001). New J. Chem. 25, 1104– 1107.
- Watanabe, M., Mataka, S. & Thiemann, T. (2005). Steroids. Submitted.